



HyCoRA

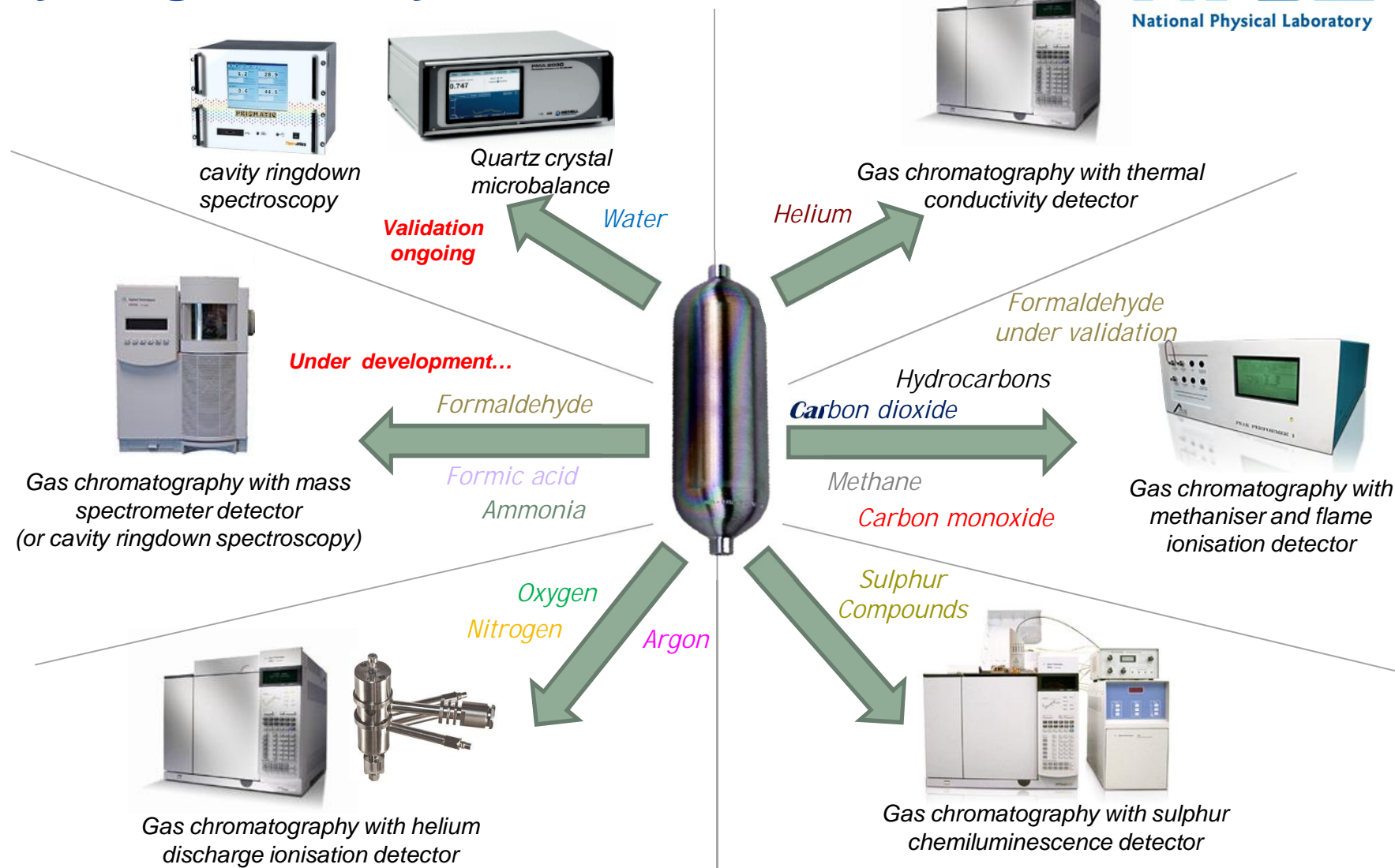
Recent development on hydrogen impurity analysis

Thomas Bacquart

2nd OEM Workshop – 9th October 2015

- **New developments on hydrogen impurities**
- New capability at NPL
- Perspective for the NPL enrichment device

Hydrogen Analysis at NPL



New development on hydrogen impurities

Reactive gases

- | | |
|--------------------------------------|--|
| • Water | (5 $\mu\text{mol/mol}$) |
| • Total hydrocarbons | (2 $\mu\text{mol/mol}$) |
| • Oxygen | (5 $\mu\text{mol/mol}$) |
| • Carbon dioxide | (2 $\mu\text{mol/mol}$) |
| • Formic acid | (0.2 $\mu\text{mol/mol}$) |
| • Carbon monoxide | (0.2 $\mu\text{mol/mol}$) |
| • Ammonia | (0.1 $\mu\text{mol/mol}$) |
| • Total halogenated compounds | (0.05 $\mu\text{mol/mol}$) |
| • Formaldehyde | (0.01 $\mu\text{mol/mol}$) |
| • Total sulphur compounds | (0.004 $\mu\text{mol/mol}$) |

Inert gases

- | | |
|------------|----------------------------|
| • Helium | (300 $\mu\text{mol/mol}$) |
| • Nitrogen | (100 $\mu\text{mol/mol}$) |
| • Argon | (100 $\mu\text{mol/mol}$) |

Non-gases

- | | |
|----------------|-----------|
| • Particulates | (1 mg/kg) |
|----------------|-----------|



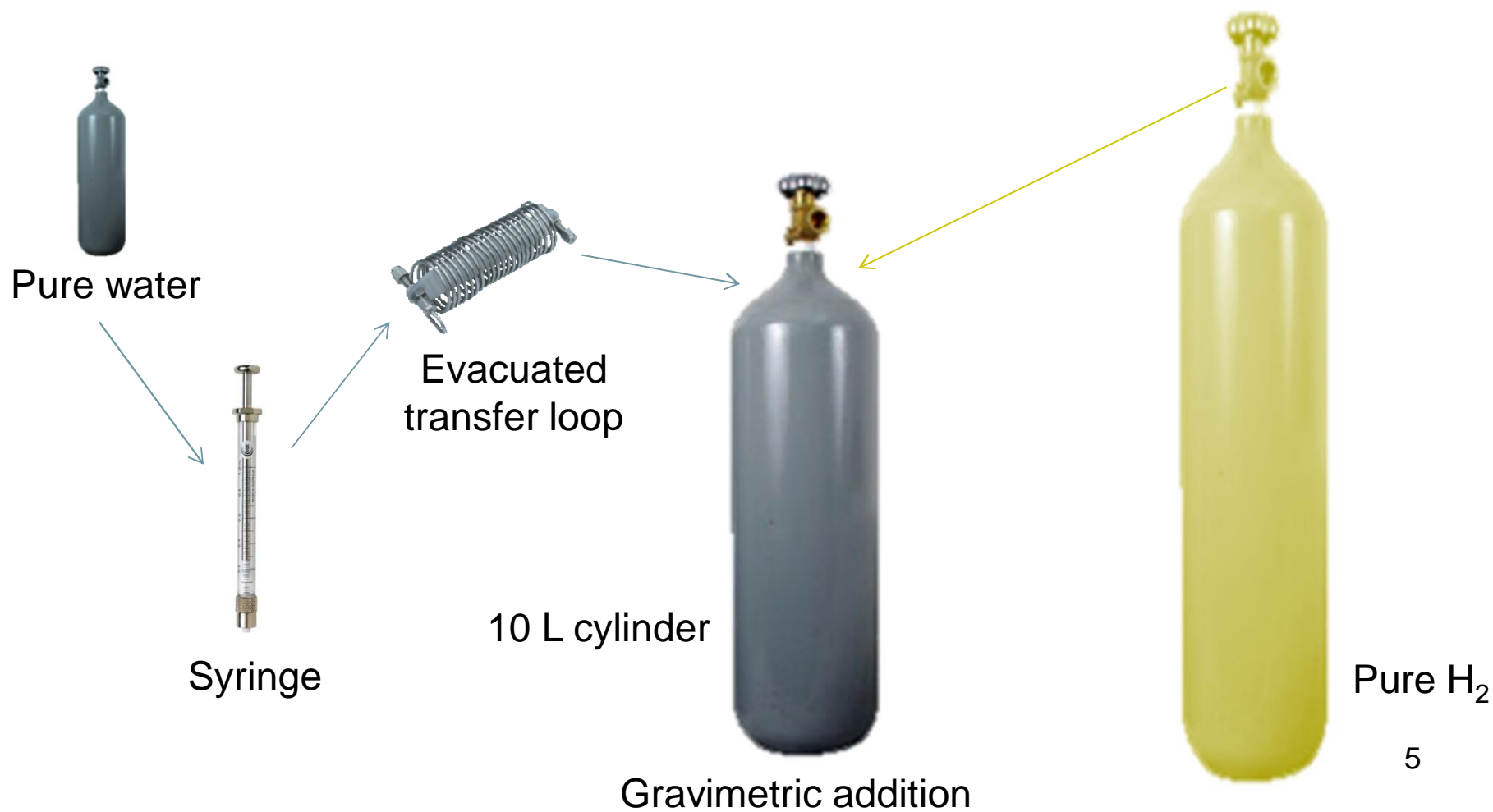
International
Organization for
Standardization

ISO 14687-2:2012

*Hydrogen fuel – Product specification –
Part 2: Proton exchange membrane
(PEM) fuel cell applications for road
vehicles*

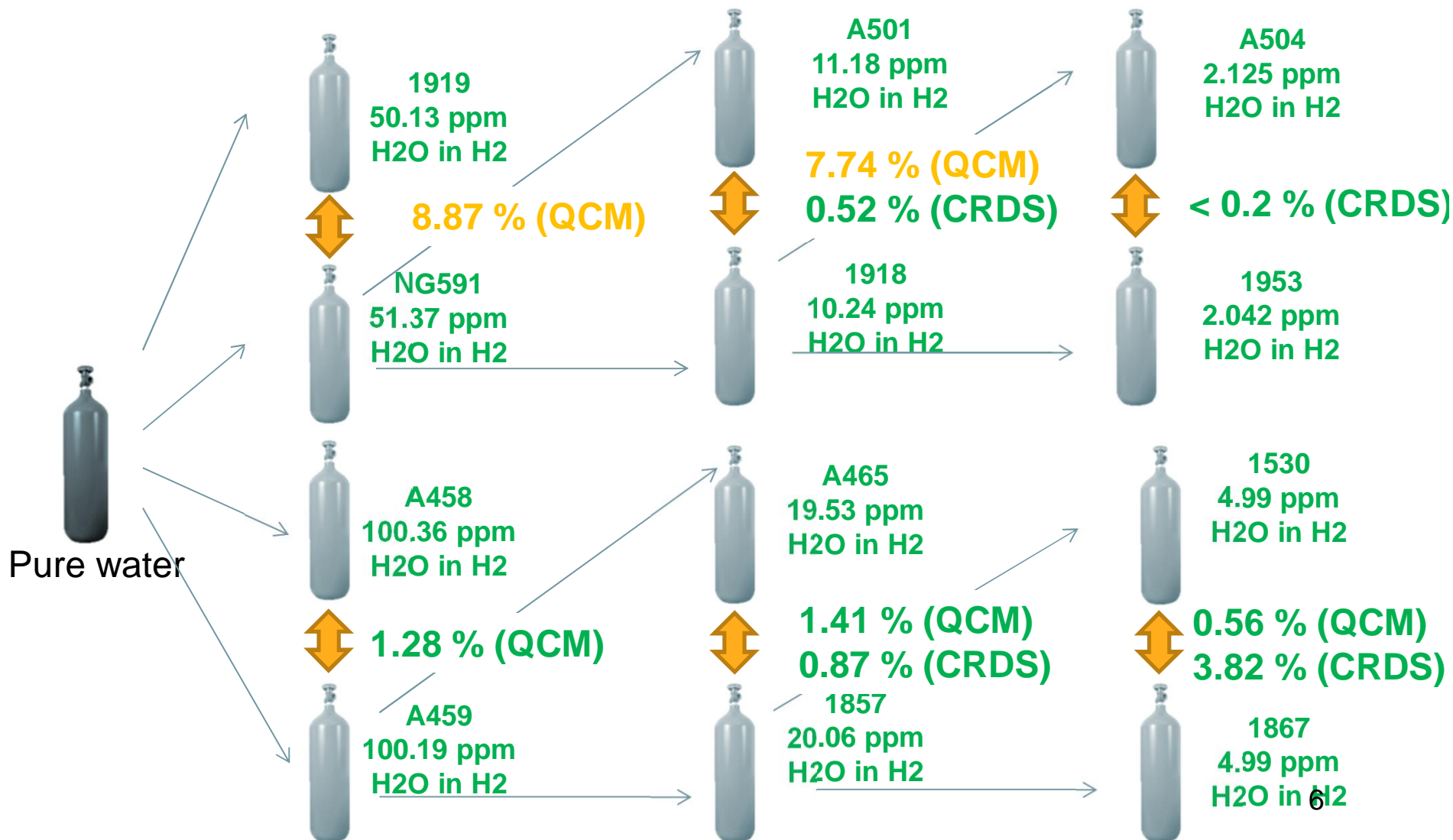
Water analysis

Water is added to the gas cylinder using a syringe and loop

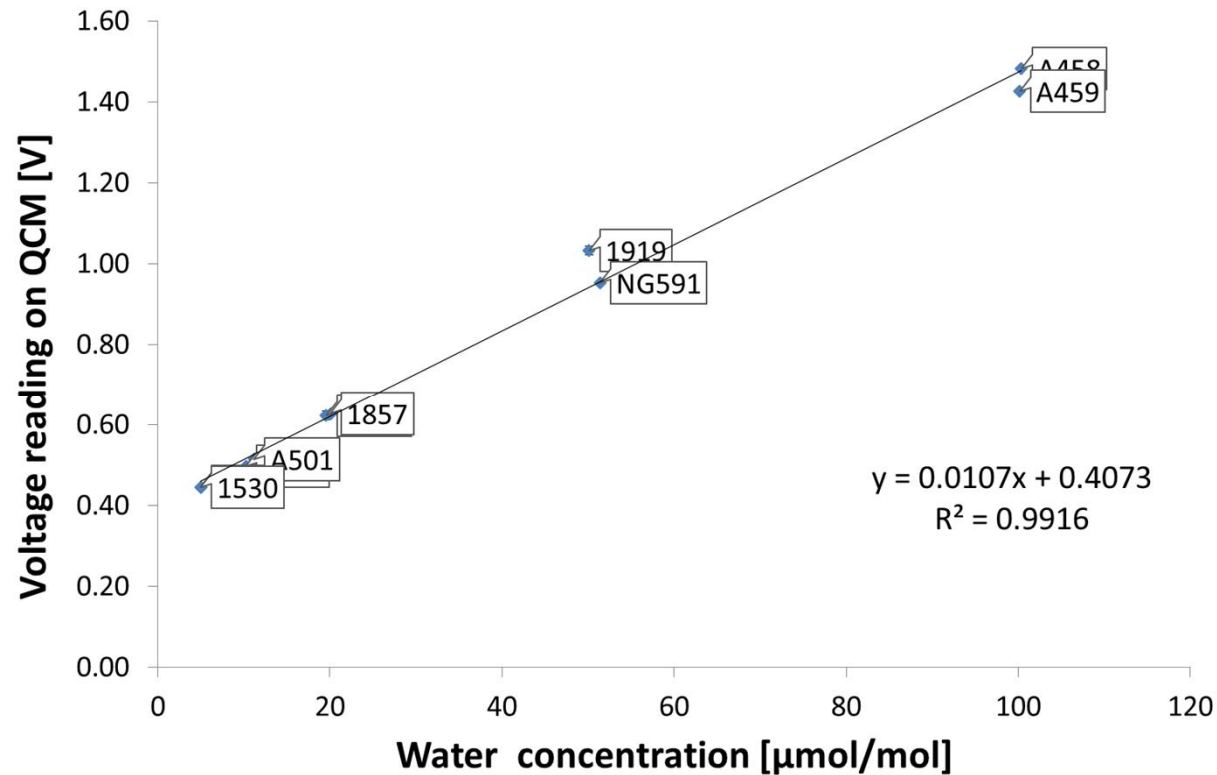


Water analysis

Primary reference materials for water content in hydrogen



Water analysis – Linearity – QCM



Acceptable linearity from 5-100 ppm water in H₂

Repeatability of slope: $\pm 30\%$ (day to day effect on instrument sensitivity)

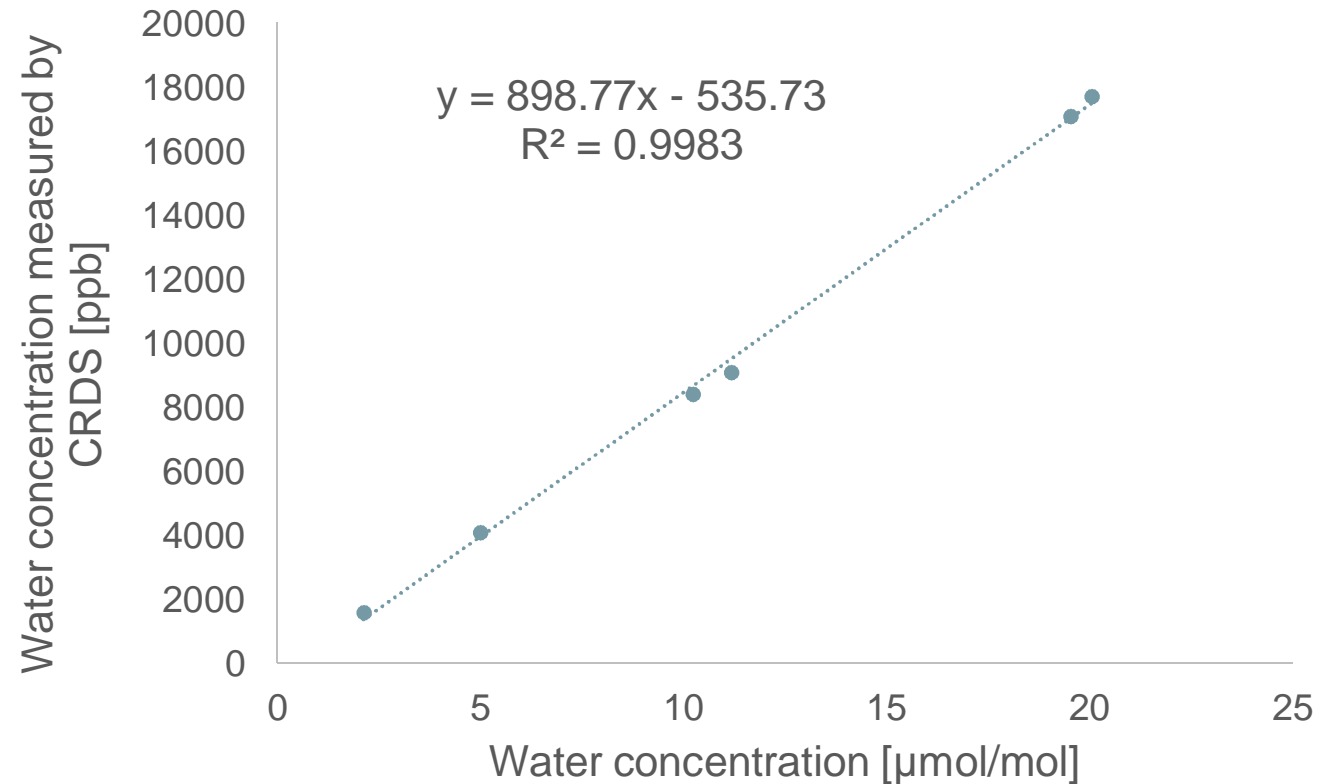
Repeatability of analysis: $< 1\%$

Intermediate precision: study ongoing

Trueness: comparison with primary method (Dew point)



Water analysis – Linearity – CRDS



Good linearity from 2-20 ppm water in H_2

Repeatability of slope: $\pm 5\%$

Repeatability of analysis: $< 2\%$

Intermediate precision: study ongoing

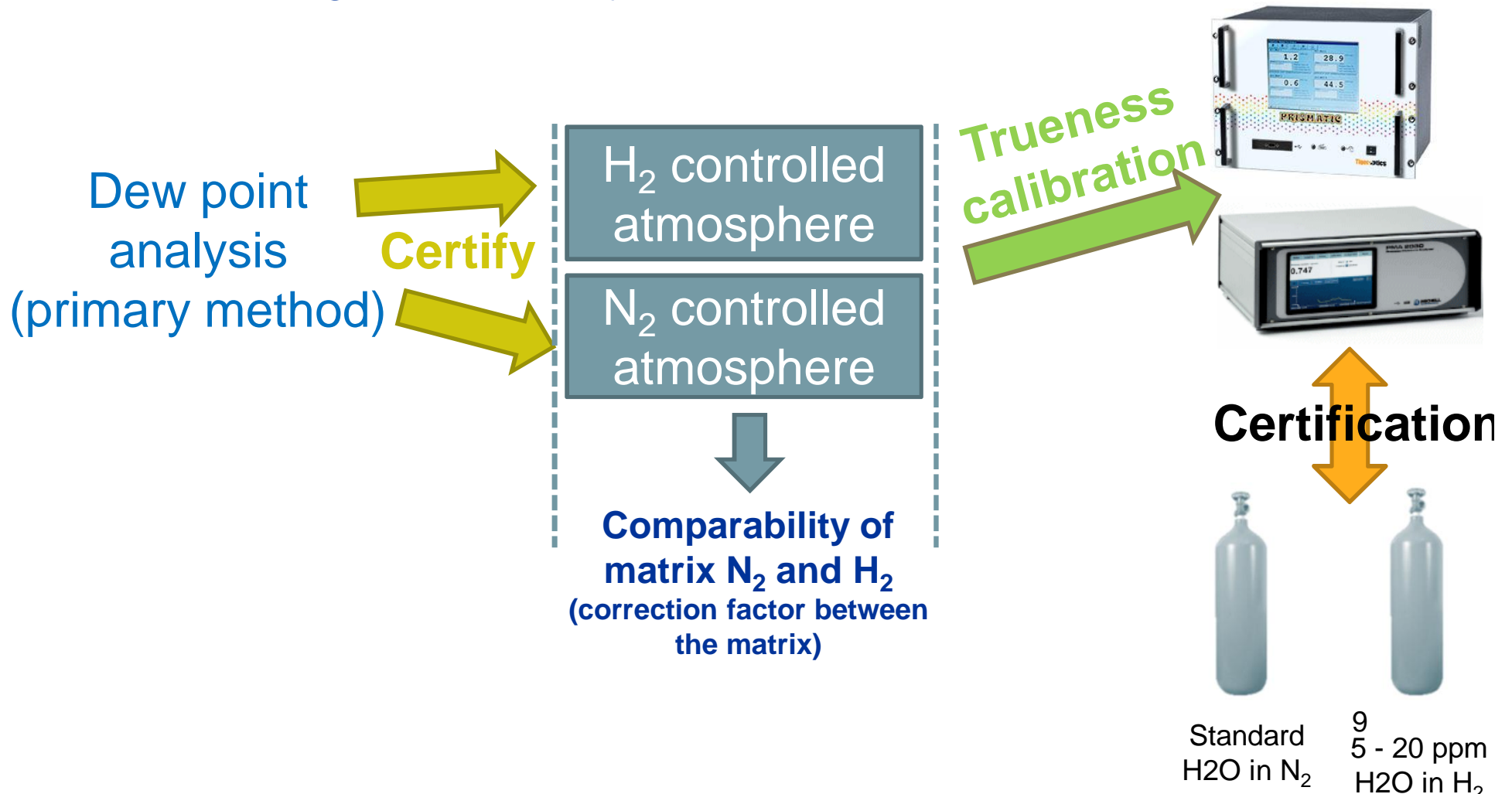
Trueness: comparison with primary method (Dew point)



Water analysis – Trueness assessment Comparability of matrix N₂ and H₂



Compare our mixtures against dynamic standards and instruments against primary standard method:



Water analysis - Summary



- CRDS
 - Linear 2-20 ppm
 - Repeatable
 - Low day-to-day effect
 - Trueness ongoing
 - Not accredited ISO17025 at NPL



- QCM
 - Linear: 5-100 ppm
 - Repeatable
 - Significant day-to-day effect
 - Trueness ongoing
 - Method accredited ISO17025 at NPL

Method extensively validated by end of October 2015

Formaldehyde analysis



NPL capability:

GC-PDHID: under development

Methaniser GC-FID: method not fully validated
(theoretical LOD ~ 10 ppb)

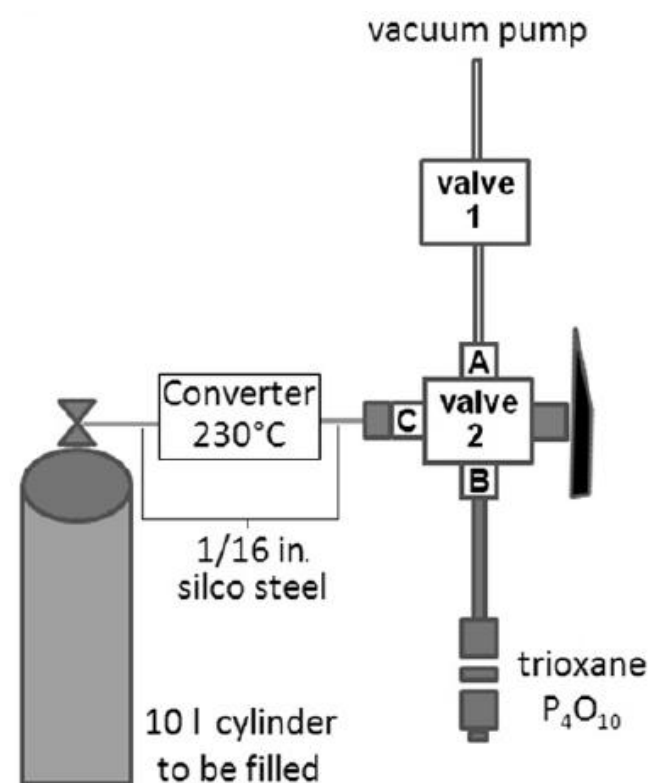
GC-MS with pre-concentration: validated by
another NPL team (LOD < 10 ppb)

FTIR: only qualitative, identification

Formaldehyde analysis

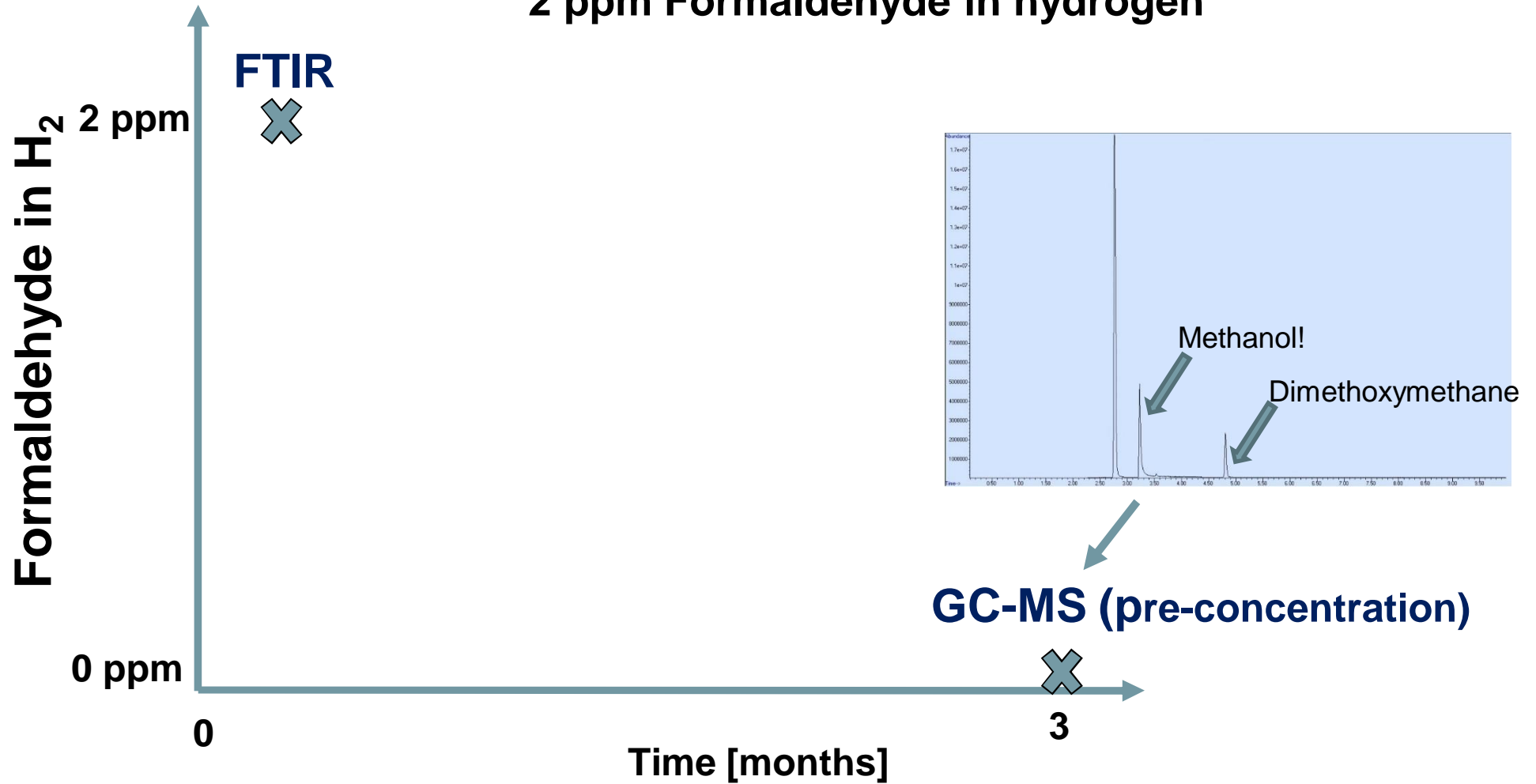
Produce and validate formaldehyde standards at 2 ppm and 10 ppm in hydrogen matrix

- Pure formaldehyde is not stable (100% formaldehyde cannot be purchased easily)
- Formaldehyde gas standards can be derived from trioxane
- System is built for producing standards in nitrogen
- **Two gas mixtures containing 2 ppm and 10 ppm of formaldehyde in hydrogen have been prepared**



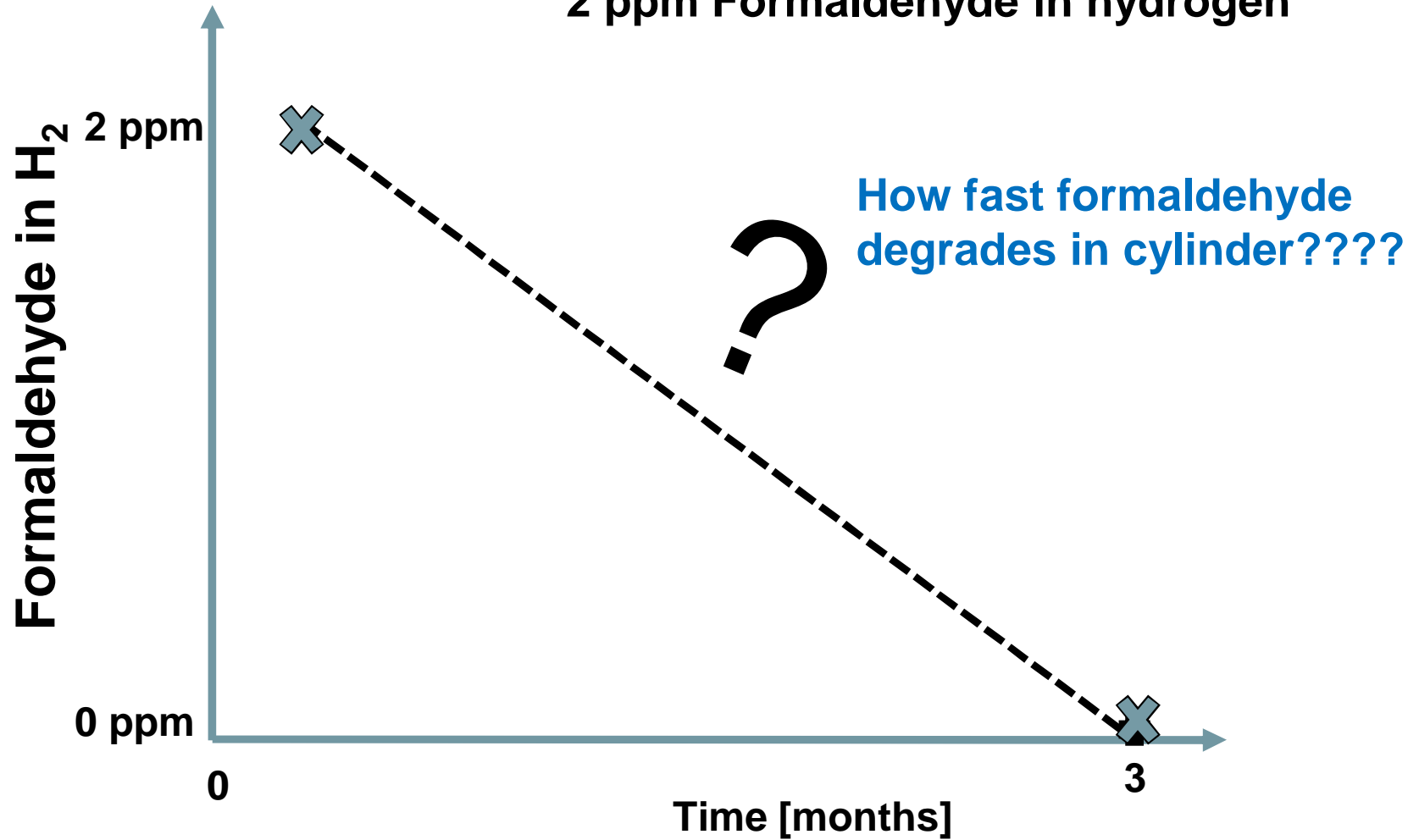
Formaldehyde analysis - 1st observation

2 ppm Formaldehyde in hydrogen



Formaldehyde analysis - 1st observation

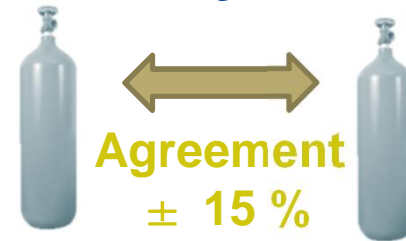
2 ppm Formaldehyde in hydrogen



Formaldehyde analysis – Stability study

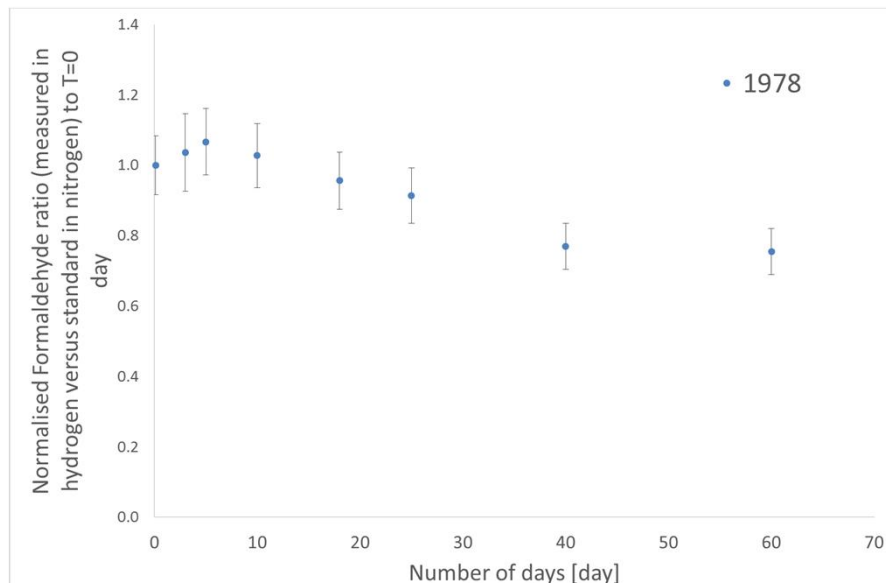
New standards:

10 ppm formaldehyde in H₂

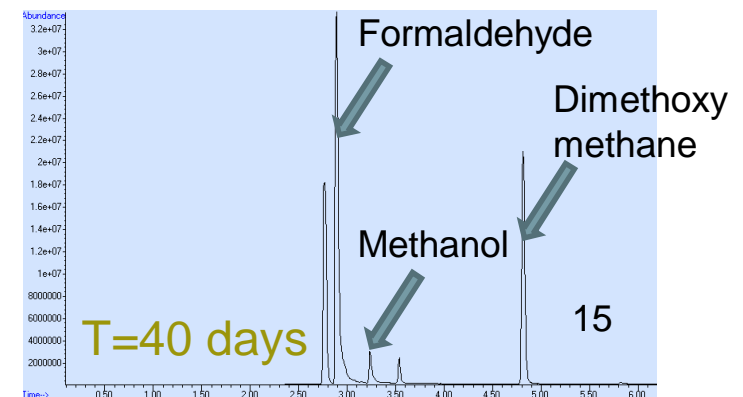
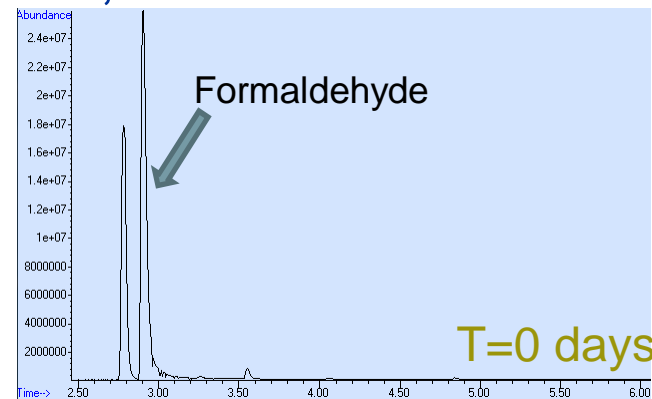


10 ppm formaldehyde
in H₂

Degradation monitoring: FTIR, GC-MS, methaniser GC-FID



Slow degradation
(loss of 10 – 20 % within 2 months)



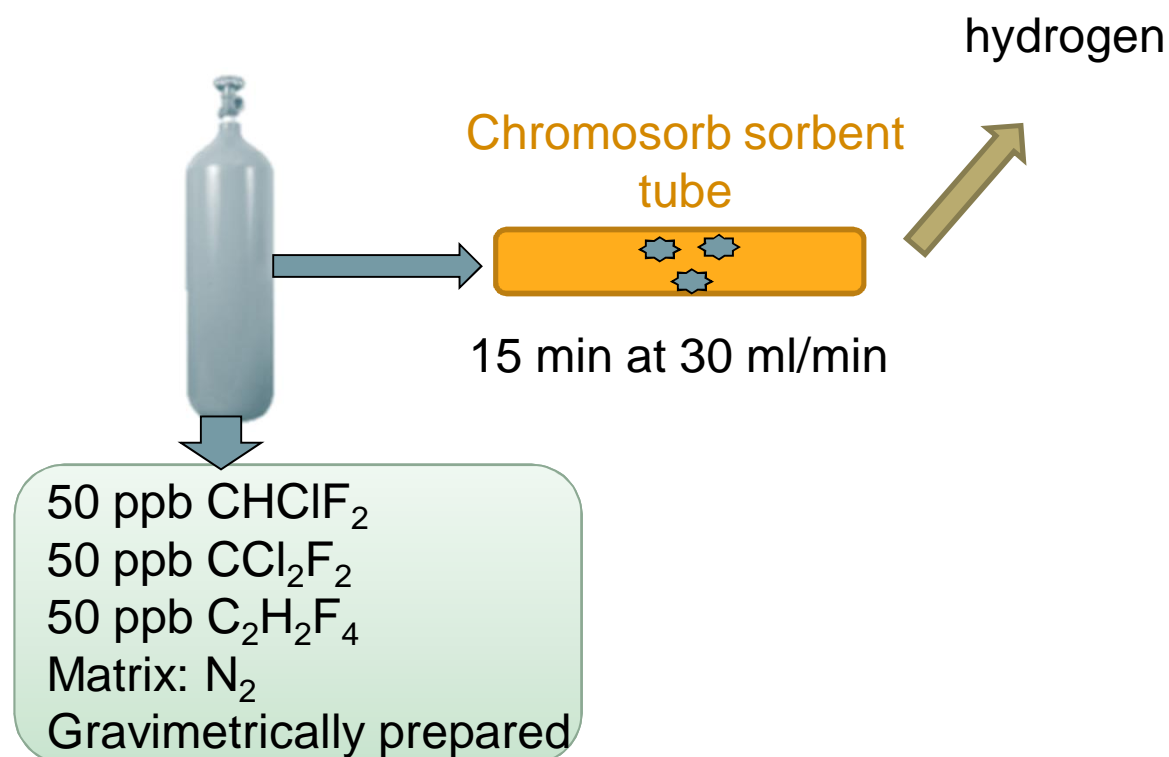
Formaldehyde analysis - Summary

- Analysis of formaldehyde in hydrogen
 - Several methods under development/validation
 - Gas standard for Formaldehyde in H₂

- Stability of formaldehyde in hydrogen
 - 2 ppm formaldehyde in H₂ → 0 % after 3 months
 - 10 ppm formaldehyde in H₂ → stable after 1 month
 - Study ongoing until complete degradation

Halogenated compounds in hydrogen

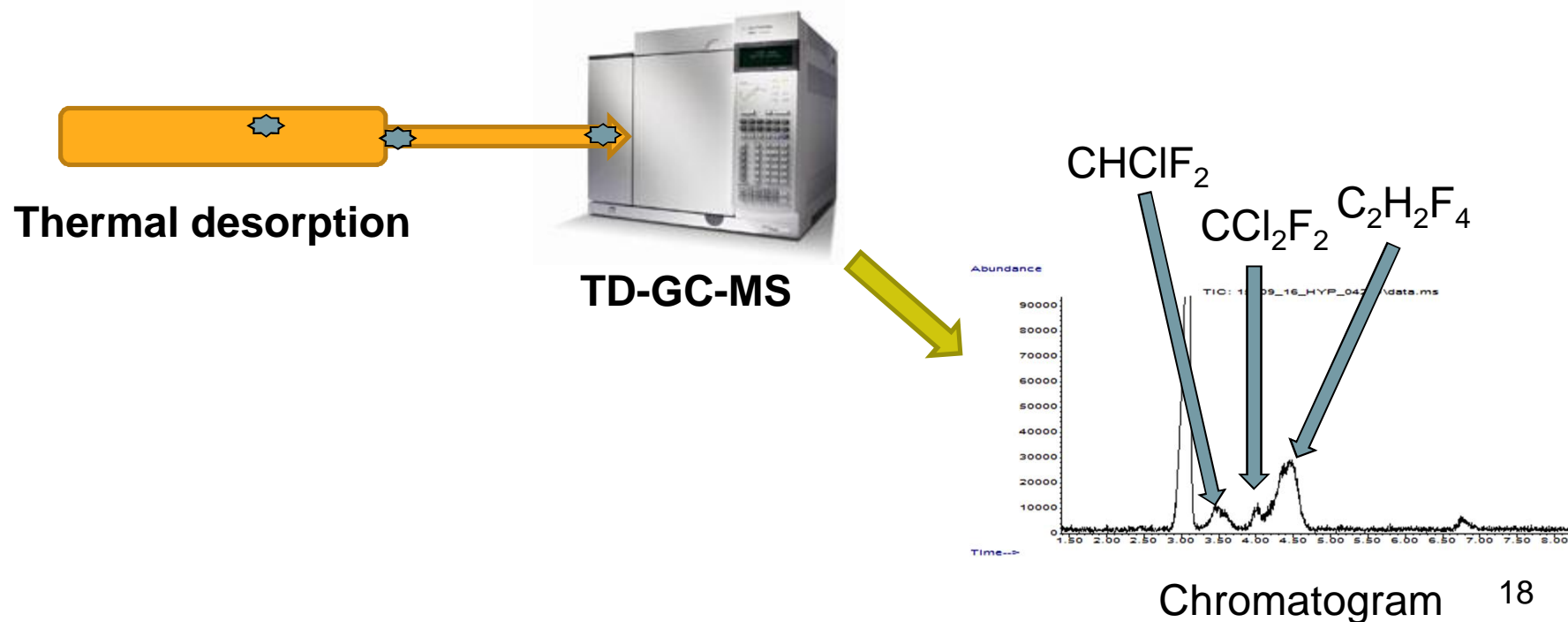
- Analysis of 3 freons gases (CHClF_2 , CCl_2F_2 , $\text{C}_2\text{H}_2\text{F}_4$) in N_2 using ThermoDesorption GC-MS (TD-GC-MS)



TD-GC-MS

Halogenated compounds in hydrogen

- Analysis of 3 freons gases (CHClF_2 , CCl_2F_2 , $\text{C}_2\text{H}_2\text{F}_4$) in N_2 using ThermoDesorption GC-MS (TD-GC-MS)



Halogenated compounds in hydrogen

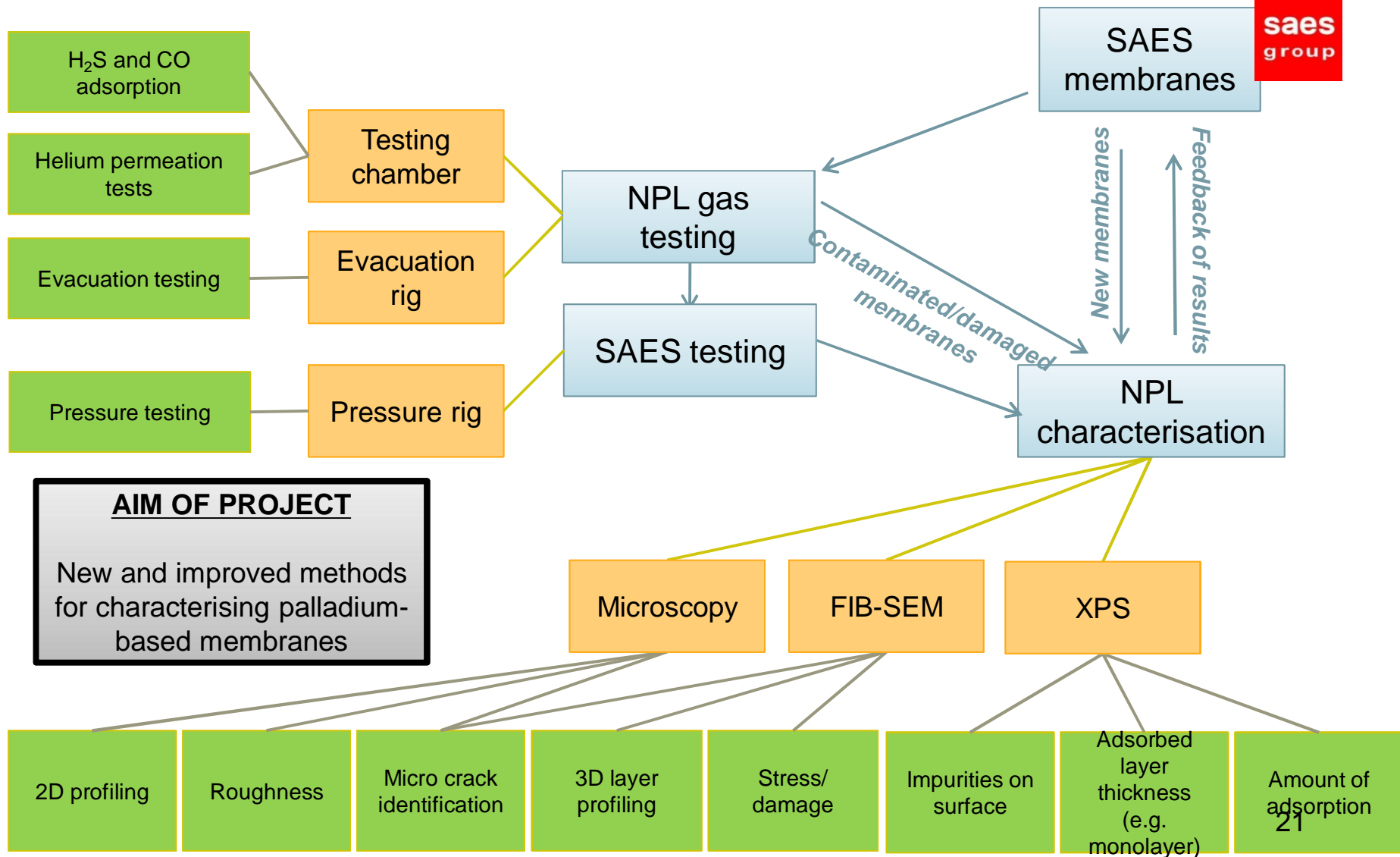
- Capability to analyse halogenated compounds at low concentration in gas

- Preparation of standard of halogenated compounds in hydrogen → to be evaluated
 - Purchasing pure compounds
 - Oxy-flame mixture
 - Stability

- Improvement of procedure → Funding required!

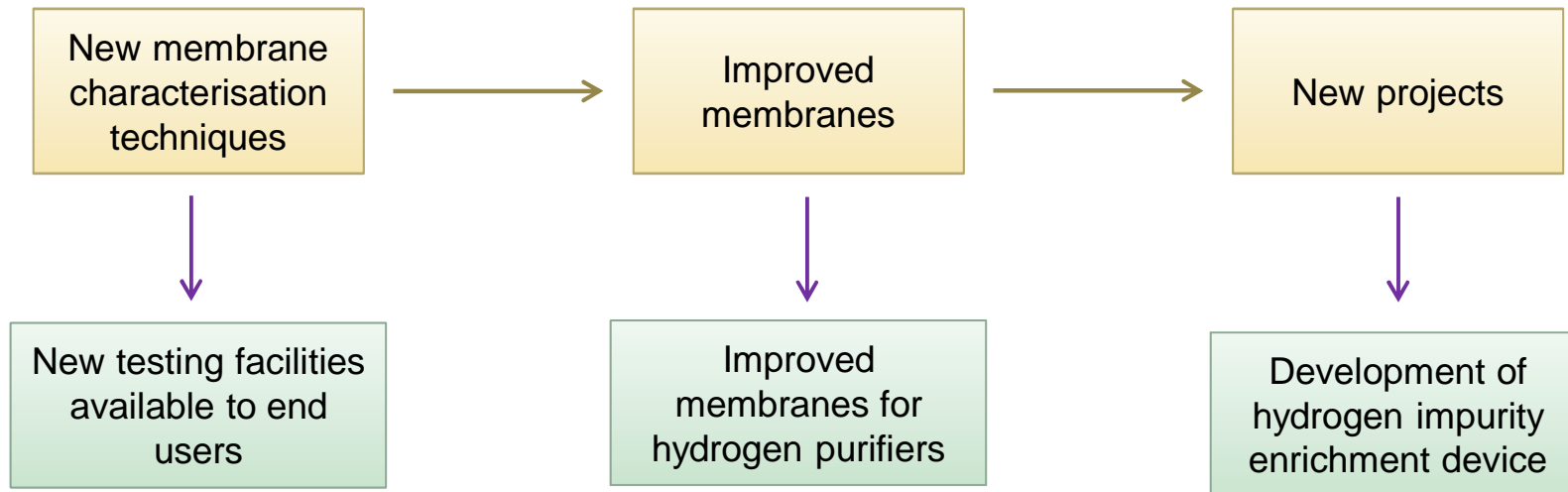
- New developments on hydrogen impurities
- **New capability at NPL**
- Perspective NPL enrichment device

Characterisation of palladium-based membranes for hydrogen purification



Characterisation of palladium-based membranes for hydrogen purification

Project outcomes



- New developments on hydrogen impurities
- New capability at NPL
- **Perspective NPL enrichment device**

Perspective NPL enrichment device



AW Automotive World

What's next for **India's passenger car** and **commercial vehicle** industries?

HOME COMMERCIAL VEHICLES ELECTRONICS MANUFACTURING OEMS & MARKETS POWERTRAIN SUPPLIERS SAFETY DIRECTORY

Home > Analysis > New device could aid design and development of next-gen FCVs, says NPL

New device could aid design and development of next-gen FCVs, says NPL

By Michael Nash | September 30, 2015

Some critics of the fuel cell vehicle (FCV) segment have suggested that until fundamental issues are resolved, the market will remain relatively small. One such issue is the degradation of fuel cells, which is caused by hydrogen fuel impurities. To combat this, the UK's National Physical Laboratory (NPL) has invented a hydrogen

Find out at Automotive Megatrends India in Pune on December 2-3.

MEGATRENDS CONFERENCES

- 2-3 DEC Automotive Megatrends India 2015 | Hyatt Regency, Pune
- 15 MAR Connected Car Detroit by Automotive Megatrends



Advanced Engineering UK
4 & 5 November 2015 - NEC Birmingham

The UK's largest annual advanced engineering event

Automotive ENGINEER

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NPL development to speed up introduction of hydrogen cars

Improved detection of impurities in fuel

Liz Wells 16 September 2015 in Technology



The diagram illustrates the process of measuring and removing impurities from hydrogen fuel. It starts with a 'hydrogen sample' containing 'Pure tracer gas (krypton)'. This is mixed with 'Krypton' to create an 'Enriched hydrogen mixture'. This mixture then undergoes 'Enrichment' to produce 'Measure impurities'. The final step is 'Measure krypton after enrichment'.

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THIS ISSUE

AE rounds up of the most important vehicle designs and concepts seen at the IAA Frankfurt motor show...

INTERNAL COMBUSTION ENGINES

2-3 December 2015

- Private partner needed for the commercialisation of the device

Conclusion

- **New development on hydrogen impurities**
 - Water analysis almost extensively validated and traceable to SI
 - Formaldehyde in Hydrogen: 1 month stability ensured
 - Halogenated analysis:
 - Organo-halogenated compounds analysed by TD-GC-MS
 - Investment needed to continue further

- **New capability at NPL**

New service: Characterisation of palladium-based membranes for hydrogen purification

- **Perspective NPL enrichment device**
 - Private laboratory partner needed to commercialise the device

Progress in other projects...

- Report of analysis for impurities in hydrogen under UKAS ISO17025 accreditation in process
- New development ongoing for ammonia, formaldehyde (GC-MS), formic acid and formaldehyde (GC-PDHID) new binary standards in hydrogen in preparation
- Participation in project proposal EMPIR co-normative call – Project proposal called Hydrogen

Acknowledgement



Hydrogen Quality
control Innovate UK

Innovate UK



Gas and Particulate Metrology group
Humidity group

Thanks,

Questions?